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Characterization of intraligand charge transfer transitions

Abstract

 $Pd(qol)_2$, $Pt(qol)_2$ and $Pt(qtl)_2$ are investigated in n-octane Shpol'skii matrices at T = 1.2 K. For the first time, it is possible to obtain highly resolved phosphorescence and triplet as well as singlet excitation spectra. The corresponding excited states are assigned to intraligand charge transfer (ILCT) transitions. Their characteristic electronic and vibrational properties are studied in detail.

Keywords: Shpol'skii spectroscopy; Intraligand charge transfer; Platinum metal compounds; Spin polarization

Coordination compounds of the ligands 8-pinelinelate NO (col-) and 8 quinelinethiclete N,S (qti) exhibit a series of interesting properties which are determined by the still largely unexplored type of intraligand charge transfer (ILCT) transitions. For example, recently, Al(qol)₃ has attracted much interest, since it plays an important role in the development of 'organic' white light emitting LEDs for future flat panel displays (e.g. see

solution at I = 270 K [2], and then use as photosensitizers for solar energy conversion was proposed [3]. For a deeper understanding of such devices a detailed characterization of the lowest

spectra of the title compounds are obtained than hitherto known (af Fig. 1(a) with 1(b)). The phase phorescence spectra recorded represent a superposition of different emitting sites. For the dominant sites, the energy positions of the triplet origins are listed in Table 1. With selective detection at these origins, site-selected highly resolved singlet excitation spectra are obtained. Due to the energy positions, oscillator strengths, and room temperature phorontion appears published in Ref. [23] these

These transitions are characterized by a charge transfer from the lone pair p-orbital at the oxygen (or sulfur) to the lowest π^* -orbital possessing more

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estimate the upper limit of the respective intersystem

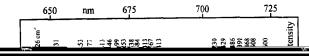
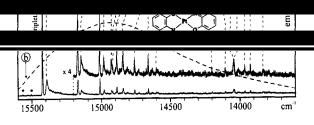


Table 1
Photophysical Data of the Lowest Excited Singlet (¹ILCT) and Triplet (³ILCT) States of Pd(qol₂), Pt(qol)₂ and Pt(qtl)₂^a



Singlet excitation

Strong FC vibr. (cm ⁻¹) Huang-Rhys factor S ^c	d d	253/413 ≈1	242/371 ≈2
Triplet emission			
$0-0^{b} (cm^{-1})$	16090	15426	13 158
FWHM ^b (cm ⁻¹)	≈3	≈3	≈3
-1:			

18 /6 / cm⁻¹ (552.85 nm, TLC 1 origin). Spectral resolution: 0.6 cm⁻¹. Vibrational satellites are specified relative to the electronic origin at 15 426 cm⁻¹. The asterisks indicate residual intensities of origins of different sites. For experimental details see Ref.

	~ 30.10		
$T_{\rm av}^{\rm g}({\rm K})$	≈5	20	≈60
$\tau_{av}^{h}(\mu s)$	160	10	1.4
Strong FC vibr. (cm ⁻¹)	236/407	253/413	241/361
			o =

crossing rate. For example, for $Pt(qol)_2$ one obtains $\approx 2 \times 10^{12} \text{ s}^{-1}$. Moreover, the highly resolved vibrational patterns found for the singlets exhibit distinct Franck-Condon (FC) progressions (for energies see Table 1) indicating shifts of the expited state potential hypersurfaces with respect to those of the ground state along the specific normal coordinates. The occurrence of these FC modes observed is connected with the charge redistributions upon excitation. A recent normal coordinate analysis for $Pt(qol)_2$ [6] showed that these in-plane vibrational modes exhibit distinct metal-nitrogen (253/

 242 cm^{-1}) and metal-oxygen/sulfur (413/371 cm⁻¹)

After excitation into the singlet origins, site-selec-

Fig. 1). The zero-field splitting (zfs) of the triplet is

smaller than 1 cm⁻¹ for all three compounds indic-

ating only a small metal d-orbital admixture to this

³ILCT (see also Ref. [7]). At T = 1.2 K, the sub-

levels emit independently (spin polarization). Thus,

the phosphorescence decay can be fitted triex-

ponentially (Table 1 and Ref. [5]). The decay be-

comes monoexponential only at relatively high

Triplet excitation

Strong FC vibr. (cm⁻¹) — 247/407 239/372

only small shifts of the triplet equilibrium positions with respect to those of the ground state.

tive triplet emission spectra resulting from the References 3ILCT are obtained (see the example given in

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- [4] H. Wiedenhofer, S. Schützenmeier, A. von Zelewsky and H. Yersin, J. Phys. Chem. 99 (1995) 13 385.
- [5] D. Donges, J.K. Nagle and H. Yersin, Inorg. Chem. (1997), in press.

levels. In contrast to the ¹ILCT spectrum, the ³ILCT shows only weak progressions, signifying

[7] H. Yersin and J. Strasser, J. Lumin., these Proceeding (ICL'96), J. Lumin. 72-74 (1997).

^an-octane Shpol'skii matrix at T = 1.2 K; concentration of compounds 10^{-5} M.

^bElectronic origin.

^cLargest Huang-Rhys factor.

eTotal zero-field splitting of the triplet sublevels.

¹ Fit components of decay curves at T = 1.2 K; I, II, III: emitting triplet sublevels.

⁸Temperature of thermalization of triplet sublevels.

^h Monoexponential decay at T_{av} .

¹Energy separation between electronic origins.